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Note

The O-polysaccharide of *Escherichia coli* O112ac has the same structure as that of *Shigella dysenteriae* type 2 but is devoid of O-acetylation: a revision of the *S. dysenteriae* type 2 O-polysaccharide structure

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Abstract—The O-antigen structure of *Shigella dysenteriae* type 2 was reinvestigated using chemical modifications along with high-resolution 2D ¹H and ¹³C NMR spectroscopy. The O-antigen was found to contain a pyruvic acid acetal, which was overlooked in an early study, and the following revised structure of the pentasaccharide repeating unit was established:

 β -D-GlcpNAc4,6(S)Pyr

1 \downarrow 3 \rightarrow 4)-α-D-GalpNAc-(1 \rightarrow 4)-α-D-Glcp-(1 \rightarrow 4)-β-D-Galp-(1 \rightarrow 3)-β-D-GalpNAc-(1 \rightarrow

where \sim 70% GlcNAc residues bear an *O*-acetyl group at position 3. The O-antigen of *Escherichia coli* O112ac was found to have the same carbohydrate structure but to lack O-acetylation. © 2008 Elsevier Ltd. All rights reserved.

Keywords: Escherichia coli; Shigella dysenteriae; O-Antigen; Bacterial polysaccharide structure; Pyruvic acid acetal

Shigella are human pathogens that cause bacilliary dysentery (shigellosis) and diarrhea, continuing to threaten public health mainly in less developed countries with conditions of poor sanitation. Although bacteria Shi-

gella are closely related to Escherichia coli and both are in fact one species, Shigella strains are put into their own genus and subgrouped into four species: Shigella boydii, Shigella dysenteriae, Shigella flexneri, and Shigella sonnei. Based on structure variation in their O-antigens, the first three species are typed into multiple serotypes. The existence of multiple O-antigen forms within a species can confer selective advantage to

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bacteria for adaptation in their econiches. More than half of the O-antigen forms of *Shigella* are shared with *E. coli* strains. 2

The structure elucidation of *Shigella* O-antigens (O-polysaccharides) has a long history and has been completed only recently (for a survey see Ref. 2). A number of the O-antigen structures, for which older methods were used, have been re-examined using modern NMR techniques and some of them revised.^{3–7}

In 1977, the structure of *S. dysenteriae* type 2 O-antigen was established⁸ (Chart 1). In the present work, this structure has been revised and close relationships of the O-antigens of *S. dysenteriae* type 2 and *E. coli* O112ac are demonstrated.

The lipopolysaccharide (LPS) of *S. dysenteriae* type 2 was isolated from dried bacterial cells by the phenolwater procedure and degraded with dilute acetic acid to give a high-molecular-mass O-polysaccharide (OPS), which was separated from low-molecular-mass materials by GPC on Sephadex G-50. The 13 C NMR spectrum indicated an irregularity of the OPS, probably owing to the presence of pyruvic acid acetal (Pyr) and *O*-acetyl groups in non-stoichiometric quantities (minor signals at δ 25.6 and 21.5). Therefore, LPS and OPS were treated with aq ammonia to give O-deacylated polymers LPSOH and DPS, respectively.

The ¹H NMR spectrum of LPSOH contained, inter alia, signals for five anomeric protons at δ 4.56–5.30, a methyl group of Pyr at δ 1.44, three N-acetyl groups at δ 2.00–2.02. The ¹³C NMR spectrum of LPSOH (Fig. 1A) showed signals for five anomeric carbons in the region δ 99.7–106.6, five hydroxymethylene groups (C-6 of hexoses) at δ 61.3–62.2 and 65.8 (data of a DEPT experiment), three nitrogen-bearing carbons (C-2 of amino sugars) at δ 50.0, 52.5, and 57.0, 17 oxygen-bearing non-anomeric sugar ring carbons in the region δ 67.1–81.8, methyl groups of Pyr (C-3) at δ 26.1 and three N-acetyl groups at δ 23.6–23.9, CO groups of Pyr (C-1) and N-acetyl groups at δ 174.9– 176.7. The position of the acetal carbon (C-2) and the carboxyl group (C-1) of Pyr was found following the correlations of Pyr H-3,C-2 and H-3,C-1 in the HMBC spectrum at δ 1.44/102.4 and 1.44/175.8, respectively.

The ¹H and ¹³C NMR spectra of LPSOH (Table 1) were assigned using 2D ¹H, ¹H COSY, TOCSY, ROESY, ¹H, ¹³C HSQC, and HMBC experiments. The ¹H, ¹H experiments revealed five spin systems for residues of β-GlcpNAc (**A**), α-GalpNAc (**B**), α-Glcp (**C**), β-Galp (**D**), and β-GalpNAc (**E**) (the sugars were denoted according to their sequence in the repeating unit shown below). The anomeric configurations of the sugar residues were ascribed based on $J_{1,2}$ coupling constant

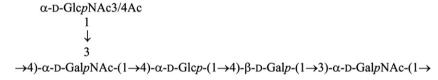


Chart 1. Structure of the O-polysaccharide of S. dysenteriae type 28.

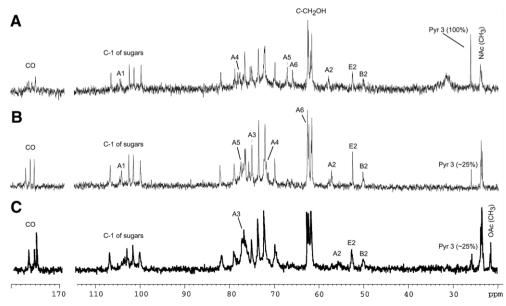


Figure 1. ¹³C NMR spectrum of LPSOH (A), DPS (B), and OPS (C) from *S. dysenteriae* type 2. Arabic numerals refer to atoms in sugar residues denoted by letters as shown in Table 1.

Table 1. ¹H and ¹³C NMR chemical shifts of LPSOH and OPS from *S. dysenteriae* type 2 (δ , ppm)

Sugar residue	Nucleus	1	2	3	4	5	6 (6a, 6b) ^a
LPSOH ^b							
β -D-Glc p NAc-(1 \rightarrow	$^{1}\mathrm{H}$	4.62	3.83	3.68	3.44	3.42	3.64; 4.02
A	¹³ C	104.2	57.0	72.3	78.1	67.1	65.8
\rightarrow 3,4)- α -D-GalpNAc-(1 \rightarrow	$^{1}\mathrm{H}$	5.30	4.14	3.86	4.29	3.96	
В	¹³ C	99.7	50.0	77.6	75.3	72.3	
\rightarrow 4)- α -D-Glc p -(1 \rightarrow	$^{1}\mathrm{H}$	4.86	3.47	3.87	3.60	4.18	
C	¹³ C	101.3	73.5	74.9	76.7	72.0	
→4)-β- D -Gal <i>p</i> -(1→	$^{1}\mathrm{H}$	4.56	3.52	3.70	3.98	3.73	
D	¹³ C	106.6	72.0	73.4	78.6	76.5	
\rightarrow 3)- β -D-GalpNAc-(1 \rightarrow	$^{1}\mathrm{H}$	4.81	4.08	3.84	4.09	3.54	
E	¹³ C	102.3	52.5	81.8	69.9	76.6	
Pyr	$^{1}\mathrm{H}$			1.44			
	¹³ C	175.8	102.4	26.1			
DPS^{c}							
β -D-Glc p NAc-(1 \rightarrow	¹ H	4.55	3.80	3.55	3.42	3.43	3.78; 3.93
A	¹³ C	104.1	57.2	75.1	71.6	77.3	62.5
\rightarrow 3,4)- α -D-GalpNAc-(1 \rightarrow	$^{1}\mathrm{H}$	5.34	4.20	3.97	4.45	4.05	3.72; 3.80
В	¹³ C	99.7	50.1	77.5	75.7	72.7	61.5
\rightarrow 4)- α -D-Glc p -(1 \rightarrow	$^{1}\mathrm{H}$	4.89	3.51	3.92	3.64	4.20	3.68; 3.88
C	¹³ C	101.4	73.4	74.9	76.8	72.0	62.2
→4)-β-D-Gal <i>p</i> -(1→	$^{1}\mathrm{H}$	4.59	3.56	3.73	4.01	3.68	
D	¹³ C	106.5	72.1	73.5	79.0	76.4	
\rightarrow 3)- β -D-Gal p NAc-(1 \rightarrow	1 H	4.96	4.14	3.88	4.14	3.74	
E	¹³ C	102.4	52.5	82.1	69.8	76.5	

^a H-6/C-6 correlations of units **B**-E in LPSOH and **D**-E in DPS are not-resolved in the region $\delta_{\rm H}/\delta_{\rm C}$ 3.65–3.87/61.3–62.2.

values (<4 Hz for α -linked and >6 Hz for β -linked monosaccharides). The absence of any 13 C NMR signals in the region δ 83–88 demonstrated the pyranose form of all sugar residues. The signals for C-3 and C-4 of unit **B**, C-4 of units **C** and **D**, and C-3 of unit **E** were shifted downfield to δ 75.3–81.8, as compared with their positions in the spectra of the corresponding non-substituted monosaccharides at δ 68.4–72.0.9 These displacements were due to glycosylation effects and defined the modes of substitution of the monosaccharides in the OPS.

A 2D ROESY experiment revealed interresidue crosspeaks between the following anomeric protons and protons at the linkage carbons: **A** H-1,**B** H-3; **B** H-1,**C** H-4; **C** H-1,**D** H-4; **D** H-1,**E** H-3; and **E** H-1,**B** H-4 at δ 4.62/3.86; 5.30/3.60; 4.86/3.98; 4.56/3.84; and 4.81/4.29, respectively (Table 1). These data are in agreement with the glycosylation pattern as determined by the ¹³C NMR chemical shift data and established the sequence of the monosaccharide residues in the O-unit.

The ROESY experiment showed also Pyr H-3,A H-4 and Pyr H-3,A H-6a correlations at δ 1.44/3.44 and 1.44/3.64, respectively, which indicated the location of the pyruvic acid acetal at positions 4 and 6 of unit A. The (S)-configuration of the acetal carbon was inferred by the Pyr C-3 chemical shift of δ 26.1 compared with published data for the corresponding (R)- and (S)-isomers. Hence, the polysaccharide chain of LPSOH has the structure shown in Chart 2.

The 1 H and 13 C (Fig. 1B) NMR spectra of the DPS showed that the majority of the repeating units lack pyruvic acid. The major series of the signals in the spectra were assigned in the same manner as for LPSOH (Table 1). A significantly lower-field position of the signals for C-4 and C-6 of unit **A** at δ 78.1 and 65.8 and a higher-field position of the signals for C-3 and C-5 of unit **A** at δ 72.3 and 67.1 in the 13 C NMR spectrum of LPSOH, as compared with their positions at δ 71.6, 62.5, 75.1, and 77.3, respectively, in the spectrum

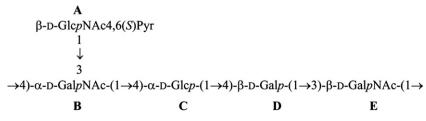


Chart 2. Structure of the polysaccharide chain of LPSOH from S. dysenteriae type 2 (this work).

^b Chemical shifts for the *N*-acetyl groups are $\delta_{\rm H}$ 2.00–2.02, $\delta_{\rm C}$ 23.6–23.9 (Me) and 174.9–176.7 (CO).

^cChemical shifts for N-acetyl groups are $\delta_{\rm H}$ 2.04–2.07, $\delta_{\rm C}$ 23.5–23.9 (Me) and 175–176.7 (CO).

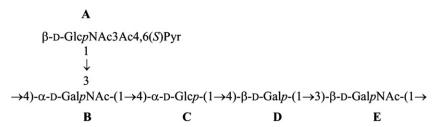


Chart 3. Structure of the O-polysaccharide of S. dysenteriae type 2 (this work).

of DPS, confirmed the location of the pyruvic acid acetal.

The position of the *O*-acetyl group was determined by a 1 H, 13 C HMQC experiment with the OPS (its 13 C NMR spectrum is shown in Fig. 1C). As compared with the 1 H, 13 C HMQC spectrum of the DPS, this showed a significant downfield displacement of $\sim 70\%$ of the A H-3,C-3 cross-peak from δ 3.55/75.1 to 5.05/76.9. This displacement is due to a deshielding effect of the *O*-acetyl group and indicated partial O-acetylation of unit A at position 3.

Therefore, based on the data obtained, it is concluded that the O-polysaccharide of S. dysenteriae type 2 has the structure shown in Chart 3. This structure differs from the published structure in the presence of a pyruvic acid acetal, which was overlooked in the earlier studies of the S. dysenteriae type 2 O-polysaccharide, 8 and in the configuration of two glycosidic linkages (units A and E are β -linked rather than α -linked).

It is known that the O-antigens of *S. dysenteriae* type 2 and *E. coli* O112ac are serologically identical. ¹¹ In order to reveal the structural basis for the serological relationship, we studied the O-antigen structure of *E. coli* O112ac. The ¹H NMR spectrum of the *E. coli* O112ac LPS showed, inter alia, a resonance at δ 1.49 and resonances from *N*-acetyl groups at $\delta \sim 2$. How-

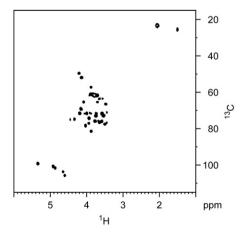


Figure 2. ¹H, ¹³C HSQC NMR spectrum of the O-polysaccharide from *E. coli* O112ac.

ever, resonances from *O*-acetyl groups at a slightly higher chemical shift were absent. The O-polysaccharide was obtained by mild acid degradation of the LPS of *E. coli* O112ac at pH 4.2. Sugar analysis by GLC of the alditol acetates after full acid hydrolysis of the polysaccharide revealed Glc, Gal, GlcN, and GalN. GLC analysis of the acetylated (*S*)-2-butyl glycosides demonstrated the D configuration of these monosaccharides.

The 1 H, 13 C HSQC NMR spectrum of the *E. coli* O112ac polysaccharide (Fig. 2) contains cross-peaks for five sugar residues and a cross-peak at $\delta_{\rm H}/\delta_{\rm C}$ 1.49/25.4 for a pyruvic acid 4,6-acetal of a hexopyranosyl residue. 12 The HSQC spectrum was in full agreement with that of the LPSOH from *S. dysenteriae* type 2, thereby confirming the structural identity of the two O-polysaccharides, except for the presence of *O*-acetyl groups in the latter (vide supra).

The O-polysaccharides of *S. dysenteriae* type 2 and *E. coli* O112ac show a structural similarity to the O-polysaccharides of another pair, *S. boydii* type 15 and *E. coli* O112ab, which are also identical structurally^{2,13} and serologically.¹¹ The latter pair differ in having an L-IdoA residue in place of D-Gal and lacking the pyruvate acetal. Nevertheless, the similarity of the O-antigens provides a serological relatedness significant enough for considering *E. coli* O112ab and O112ac as variants of the serogroup.

1. Experimental

1.1. Bacterial strains and isolation of lipopolysaccharides

S. dysenteriae type 2 (G1252) was from the Institute of Medical and Veterinary Science, Adelaide, Australia, and E. coli O112ac:K66:H⁻ (CCUG 11414) was from the Culture Collection, University of Gothenburg, Sweden. Bacteria were grown to late log phase using a fermentor under constant aeration at 37 °C and pH 7.0 as described for S. dysenteriae type 4;⁶ bacterial cells were washed and dried. Lipopolysaccharides were isolated from dried bacterial cells by the phenol–water method¹⁴ in a yield of 7% and purified as described.⁶ E. coli O112ac was grown and extracted as previously described.¹⁵

1.2. Degradation of lipopolysaccharides

Delipidation of the LPS of *S. dysenteriae* type 2 was performed by treatment with aq 2% HOAc or 0.1 M sodium acetate pH 4.2, respectively, at 100 °C for 5 h. The precipitate of lipid A was removed by centrifugation (13,000g, 20 min), and the high-molecular-mass OPS was recovered from the supernatant by fractionation by GPC on a column (56×2.6 cm) of Sephadex G-50 (S) (Amersham Biosciences, Sweden) in 0.05 M pyridinium acetate buffer pH 4.5, monitored by a differential refractometer (Knauer, Germany).

The LPS and OPS of *S. dysenteriae* type 2 were treated with aq 12.5% ammonia at 37 °C for 16 h, the solution was desalted by gel filtration on Sephadex G-50 (S) as described above to give the O-deacylated polymers LPSOH and DPS, respectively. *E. coli* O112ac was delipidated and purified as previously described.¹⁵

1.3. Chemical analyses

The OPS of *E. coli* O112ac was hydrolyzed with 2 M CF₃CO₂H (120 °C, 2 h). Monosaccharides were identified by GLC of the alditol acetates on a Hewlett–Packard 5890 Series II instrument equipped with an HP fused silica column (0.25 mm × 30 m) using a temperature program of 180 °C (1 min), a temperature increase to 210 °C (3 °C min⁻¹) and 210 °C (10 min). The absolute configuration of the monosaccharides was determined by GLC of the acetylated (*S*)-2-butyl glycosides.¹⁶

1.4. NMR spectroscopy

NMR spectra of the DPS and LPSOH of *S. dysenteriae* type 2 OPS were measured at 30 and 40 °C, respectively, using a Bruker DRX-500 MHz spectrometer. NMR spectra of the OPS of *E. coli* O112ac were recorded at 55 °C using a Bruker AVANCE 500 MHz spectrometer equipped with a 5 mm PFG triple-resonance CryoProbe. Samples were freeze-dried from 99.9% D₂O and studied as solutions in 99.96% D₂O using acetone ($\delta_{\rm H}$ 2.225) or sodium 3-trimethylsilyl-[2,2,3,3- 2 H₄]propanoate ($\delta_{\rm H}$ 0.00) as the internal reference and acetone ($\delta_{\rm C}$ 31.45) or dioxane ($\delta_{\rm C}$ 67.4) in D₂O as the external reference. Data processing was performed using vendor-supplied software.

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